ACCOUNTS ASSIGNMENT

MANAGEMENT ACCOUNTING- Do question 10, 11, 12, 13, 18, 19 and 20 of chapter 2 in Accounts notebook.

English Language

Topic- Narrative Composition

Question 1

Write a composition (in approximately 400--450 words) on any one of the following topics.

We can never know how a man feels, until we have walked in his shoes.' Narrate a personal experience, which brings out the truth of this saying.

It is said, "If at first You don't succeed, try, try again." Narrate your experience when you persisted until you achieved your goal.

POLITICAL SCIENCE

- 1. What is Dictatorship? Explain any six of its features.
- 2. Give one example each of a Authoritarian /Dictator state and one which is a Totalitarian Communist State.
- 3. Explain any six points of criticism against Dictatorship.
- 4. Briefly explain any 8 points of difference between Totalitarian State and Authoritarian State.

CHEMISTRY

- A. Give reasons for the following:
- 1. Moving down the group in oxygen family the acidic nature of its hydrides increases and thermal stability decreases.
- 2. Why boiling point of water is higher than the other hydrides of group 16?
- 3. Water is liquid but H2S is a gas. Why?
- 4. Oxygen is a diatomic gas while other members of the family are solids.
- 5. What is the mode of diluting conc. Sulphuric acid?
- 6. Why we add acid to water and not water to acid while diluting conc. Sulphuric acid?
- 7. SO3 is an acid anhydride of H2SO4 then why SO3 is mixed with H2SO4 to form oleum during the contact process?
- B. Answer the following:
- 1. Give examples of the compounds in which oxygen exhibit positive oxidation state.
- 2. Except oxygen why other members of the family show +4 and +6 valencies?
- 3. Why electron gain enthalpy of oxygen is less than that of sulphur?

- 4. Why oxygen behaves anomously different from other members of the family?
- 5. How the nature of oxides changes while moving left to right in the periodic table. Explain with an example.
- 6. Discuss the preparation of ozone using Siemen's ozoniser.
- 7. Write the reactions for ozone showing its oxidizing nature with lead sulphide, potassium iodide and mercury.
- 8. Mention various allotropic forms of sulphur.
- 9. Write the chemical reaction with NaOH, Cl2 and KMnO4. Also draw the structure of SO2.