## POLITICAL SCIENCE ASSIGNMENT

- 1. On what basis are the states classified as Parliamentary and Presidential?
- 2. What do you mean by Parliamentary form of Government?
- 3. Define Political Homogeneity in a Parliamentary form of government.
- 4. What is meant by Cabinet Dictatorship? Which type of government would you associate with it?
- 4. Discuss any 6 characteristic features of Parliamentary form of government.

## **ACCOUNTS ASSIGNMENT**

MANAGEMENT ACCOUNTING- Do question 59,61,67,68,70,73,74,75,77 and 78 of chapter 4 in Accounts notebook.

## **CHEMISTRY ASSIGNMENT**

Nomenclature and Isomerism in Inorganic Complexes:

A. Give the structural formula of the following compounds:

- 1. potassium tetrafluoridoborate(III)
- 2. hexaamminecobalt(III) chloride
- 3. pentacarbonyliron(0)
- 4. hexacyanidoferrate(II)ion
- 5. diamminesilver(I)ion
- B. Answer the following:
- 1. Write all the geometrical isomers of [Pt(NH<sub>3</sub>)(Br)(Cl)(py)] and how many of these will exhibit optical isomers?
- 2. Name the type of isomerism exhibited by the following pairs of compounds:
- i)  $[Co(ONO)(NH_3)_5]^{+2}$  and  $[Co(NO_2)(NH_3)_5]^{+2}$
- ii)  $[Cr(H_2O)_4Cl_2]Cl.2H_2O$  and  $[Cr(H_2O)_5Cl]Cl_2.H_2O$
- iii)  $[Co(NH_3)_6][Cr(CN)_6]$  and  $[Cr(NH_3)_6][Co(CN)_6]$
- 3. There are two isomers

 $[Co(NH_3)_5(NO_3)]SO_4$  and  $[Co(NH_3)_5(SO_4)]NO_3$  which complex will give white precipitates with barium chloride. Also write the type of isomerism.

4. Write cis and trans isomer for  $[CoCl_2(en)_2]^+$ . Which one is optically active and why?